

**ChE 323 Mass Transfer  
Spring 2005**

**Homework # 3  
Due Friday April 29**  
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Consider Example 3.2.9 of your text (pp. 67-68). Use MATLAB to obtain the full numerical solution to the boundary value problem described by equations 3.2.195-3.2.200. Note that the boundary value solver `bvp4c` can only solve systems of first-order equations. Therefore you must recast this problem as a system of first-order equations. Use the following values for the constants in the problem:

$$\begin{aligned}D_1 &= 1.99\text{e-}5 \text{ m}^2/\text{s} \\D_2 &= 4.55\text{e-}6 \text{ m}^2/\text{s} \\k &= 3.33\text{e-}3 \text{ m}^3/\text{mol}\cdot\text{s} \\C_{10} &= 1.2 \text{ mol}/\text{m}^3 \\C_{2L} &= 0.8 \text{ mol}/\text{m}^3 \\L &= 0.4 \text{ m}\end{aligned}$$

- Submit a plot of the concentration profiles of species 1 & 2 in the film (on the same plot). Do they make sense qualitatively?
- Calculate the flux of species 1 at  $z = L$ .
- How big must  $k$  become before the approximation of equations 3.2.207 – 3.2.209 becomes valid?
- How does the result of part (a) change if  $D_1 = 4.55\text{e-}7 \text{ m}^2/\text{s}$ ? Is this result expected?